the moon are  $8.2 \times 10^6$  cm for water and  $3.4 \times 10^6$  cm for carbon dioxide. These gases thus expand into effective volumes of  $3.1 \times 10^{24}$  cm<sup>3</sup> and  $1.3 \times 10^{24}$  cm<sup>3</sup>, respectively. To within an order of magnitude, the pressure rises to be expected due to ejection of this amount of gas are  $2 \times 10^{-13}$  torr for water and  $5 \times 10^{-13}$  torr for carbon dioxide. Pressures of this magnitude are measurable with commercially available equipment.

Simple estimates of typical escape times for these gases indicate that they will remain for at least several thousand years. We may then expect to modify the total lunar environment irreversibly, and only partly predictably, each time a rocket lands there. Only if the natural background pressures of water and carbon dioxide are several orders of magnitude larger than the above values will our perturbations of these quantities be unimportant.

JOHN O. STONER JR University of Arizona

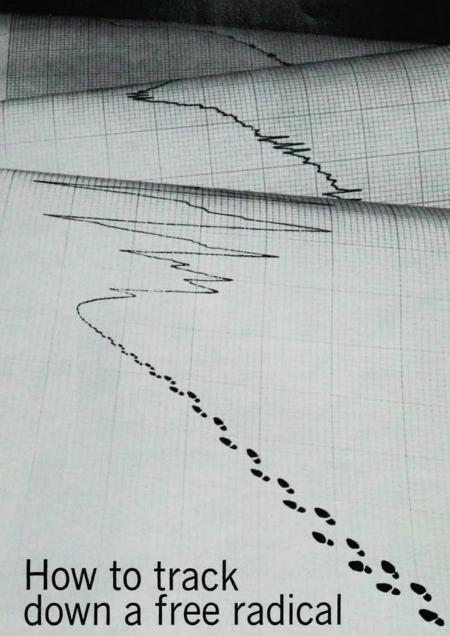
## Emily Wolf and register

I enjoyed the article "The National Register Looks at Manpower" in the October Physics Today. In one statement, though, it is in error.

At the request of Henry A. Barton, then director, and Wallace Waterfall, then as now secretary of the American Institute of Physics, I organized the register in November 1953. I employed Sylvia Barisch, your senior author, in March 1954 as one of my part-time coders. I remained in charge of the register until 1960, when it was transferred to the newly formed Education and Manpower Division. Mrs Barisch had been named supervisor in May 1959.

EMILY WOLF
American Institute of Physics

Correction: The editors apologize for two typographical errors in Don B. Lichtenberg's October review of Paradoxes in the Theory of Relativity by Yakov P. Terletskii. The word "comparable" was substituted in the last sentence, which should have read, "... the theory of relativity is compatible with dialectic materialism." The first equation in the fourth paragraph should have read  $\gamma = 1/(1 - u^2/c^2)^{\frac{1}{2}}$ .



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