

zones, which has proved so useful in the theory and interpretation of electron energy bands in metals. Essentially, it is a study of the wave properties of the solutions of Schrödinger's equation for periodic potential fields, which depend on the degrees of symmetry (group properties) associated with various crystallographic forms. Therefore, the theory and its applications for the determination and classification of energy states in crystals is based on the single-electron model, which, in spite of its shortcomings, has been very successful in the understanding and interpretation of the experimental data.

The author begins with a discussion of the solution of the one-dimensional Schrödinger equation in a periodic field, the Bloch functions, a detailed account of Brillouin zones for various crystallographic groups, and symmetry properties of the wave functions of these groups. This is followed by a detailed analysis of group operations and their reduction to regular representation for point groups, and tables of characters of point groups, and the classification of electronic states associated with different crystallographic groups. Of particular importance is the discussion and analysis of electron states in space groups containing glide and screw displacements with application to closely packed structures (especially the diamond) and the extensions of the theory to large zones. The last two chapters contain a treatment of the determination of the wave functions and energy bands by using such methods as the linear combination of atomic orbitals, the method of superposition of plane waves, and the so-called cellular method. The latter procedure has yielded successful results when the degree of symmetry is high. A discussion of the effects on the structure of the states brought about by the coupling of the spin with the angular momentum of the electron is also included. The references accompanying each chapter, although not exhaustive, provide the reader with sources on recent developments.

Theorie des Aufbaues der Materie. By Friedrich Hund. 313 pp. B. G. Teubner Verlagsgesellschaft mbH, Stuttgart, 1961. DM 37.60. *Reviewed by R. Landshoff, Lockheed Missiles and Space Company.*

INTERATOMIC forces play an important role in determining the structure of matter; they are responsible for the binding of atoms into molecules and for their association in crystals, as well as less regular solids or liquids. The long-range electrostatic forces between ions are a familiar concept of classical theory; most other forces between atoms or molecules can only be explained with quantum theory.

This book introduces the reader to the most important ideas which give an understanding of how the various types of association come about. After a discussion of condensation, the book turns to crystal lattices and to the factors which influence the type of lattice formed by specific atoms and ions.

After a brief introduction to quantum theory, it is

shown how one can explain interatomic forces. The formalism is developed for the diatomic molecule and then extended to more complicated types of chemical binding. An important aspect of this theory is the existence of symmetries and their utilization by means of group theory. The latter part of the book deals with properties of solids, particularly those which are influenced by the grouping of electronic energy levels into bands. The Bardeen-Cooper-Schrieffer theory of superconductivity has unfortunately been omitted.

The book emphasizes a qualitative rather than a quantitative understanding and steers away from detail which might detract from the main argument. Such a program has an obvious appeal, but in the process of eliminating detail, the book occasionally goes a bit too far. For the trained physicist, who can reproduce missing links, this presents no difficulty, but the novice may require a helping hand or additional source material. With such help, the presentation will indeed give him a clear understanding of the basic concepts which are used to explain the structure of matter.



Enrico Fermi

Notes on Quantum Mechanics. A Course Given by Enrico Fermi. 171 pp. U. of Chicago Press, Chicago, Ill., 1961. Paperbound \$1.50. *Reviewed by Kamal K. Seth, Duke University.*

ALMOST all of the creators of quantum mechanics have at one time or another written a book on the subject, and most of these books, for example Heisenberg's or Dirac's, have become classics. They vary in their approaches as well as in level of presentation. Consequently, a teacher of quantum mechanics usually has to formulate his own approach to the subject. Several excellent texts in the field also exist (for example Schiff's, which has in its own right become a classic)